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1. A group of atoms united by ionic bonds is called a molecule. ____ 2. A covalent bond is formed by a shared pair of electrons

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REVIEW and REINFORCEMENT Section Ionic Bonds

Ionic solids exhibit a crystalline structure and tend to be rigid and brittle; they also tend to have high melting and boiling points, which suggests that ionic bonds are very strong. Ionic solids are also poor conductors of electricity for the same reason—the strength of ionic bonds prevents ions from moving freely in the solid state.

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7-2 Review and Reinforcement Covalent Bonding Complete each statement by filling in the correct term. If the statement is true, write "true." If it is false, change the underlined word or words to make it true. ... Ionic compounds that absorb water into their solid structures form When naming an ionic substance, the name of the

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Bonding Review Station #2: 1. Ionic bonds are formed by a TRANSFER of electrons. 2. Anions are attracted to a POSITIVE charge. 3. Melting points of ionic compounds are typically (higher/lower) than those of covalent compounds. 4. What does it mean to say that an ionic compound is "brittle"? Brittle means to break apart easily and in smaller

ANSWER KEY Bonding Review Station #1: What charge does a ...

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