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Chapter 8 Concepts of Chemical Bonding . Chemical Bonds Three types: - Ionic Electrostatic attraction between ions Covalent Sharing of electrons Metallic Metal atoms bonded to several other atoms Covalent Bonding • A bond where electrons from each atom are shared

Section 8.2 - The Nature of Covalent Bonding In ionic bonding, atoms transfer electrons to achieve noble gas configuration. In covalent bonding, atoms share electrons to achieve noble gas configuration. Most atoms share electrons until they have a total of 8 valence electrons (octet rule).

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